SET

A

THIRD SEMESTER ENVIRONMENTAL POLLUTION & MANAGEMENT MSC - 306A [SPECIAL REPEAT]

M.Sc. CHEMISTRY

USE OMR FOR OBJECTIVE PART

Duration: 3 hrs.

Full Marks: 70

Objective]

Time: 30 min.

Marks: 20

Choose the correct answer from the following:

1X20=20

- 1. Nitrification is important in nature, because
 - a. Plants absorb nitrogen in the form of ammonium salts
 - c. Both a and b

- Plants absorb nitrogen in the form of nitrate
- d. None of these
- The hazardous chemical can enter the human body through which of the following routes
 - a. Ingestion
 - c. Eye contact

- b. Inhalation
- d. All the above
- 3. What does PPE stand for?
 - a. Professional Personal Equipmement
 - c. Personal Protective Equipment
 - d. Properly Protected Employee
- 4. Nylon threads are made of
 - a. Polyester polymer
 - c. Polyethylene polymer
- b. Polyamide polymer

b. Protective Practical Equipment

d. Polyvinyl polymer

5. This sign represents



- a. Corrosion
- c. Toxic

- b. Flammable
- d. Explosive
- 6. What is the main purpose of hazard identification?
 - a. To minimise the effect of a consequence
- b. For better risk management
- c. To characterize adverse effect of toxins
- d. To reduce probability of occurrence
- 7. Denitrification step involves
 - a. Reduction of nitrate to nitrogen gas
- b. Reduction of ammonia to nitrogen gas
- c. Oxidation of nitrogen gas to ammonia
- d. None of the above

ng is • Reuse • All of the above	
UV-light radiation All of the above	
Flow of energy None of the above	
nts of soil? Sublimation Loams	
To maximize surface run- Not storing surplus rainw	the soil
ter and topsoil from one place. Soil degradation Soil pollution	What is called for the movement of surface litt another? a. Soil submerge b
se of Stored water Chemical fertilizers	
o. Two I. Four	
particulate pollutants? De Radon Le Ethylene	
sible for acid rain? Carbon dioxide Nitrogen dioxide	
	The WHO provisional guideline value of fluora. 1.5 mg/L b c. 10 g/L d
o. Carbon dioxide d. Nitrogen dioxide ride in drinking water is o. 2.5 mg/L	a. Sulphur c. Sulphur dioxide d The WHO provisional guideline value of fluora. 1.5 mg/L b

USTM/COE/R-01

19. Eutrophication of water bodies leading to killing of fishes is mainly due to nonavailability of a. Food b. light c. Oxygen d. Essential minerals 20. Which element is responsible for Eutrophication? b. phosphorus a. Nitrogen c. Both a and b d. None of the above [Descriptive] Time: 2 hrs. 30 mins. Marks:50 [Answer question no.1 & any four (4) from the rest] 1. a. Explain the mechanism of petroleum hydrocarbon degradation. 5 What are the factors that influence petroleum hydrocarbon degradation? b. What is meant by Industrial waste? What are the different types 5 of industrial waste? Explain briefly. 2. a. What do you mean by wastewater treatment? What are the roles 1+3=4 of atmosphere? b. What are the main cycles of nature? Show key steps of nitrogen 6 cycle with proper diagram. a. What are the key steps involved in carbon cycle? Show with 5 proper diagram. b. Classify solid wastes based on their sources of origin and on their physical nature. What do you mean by solid waste 2+1+2 management? What are the methods for treatment of solid

waste?

4.	a.	Write short notes on (i) Detergents (ii) Dye.	2+2=4
	b.	What do you mean by synthetic polymers? Write three different types of synthetic polymers and their uses.	(
5.	a.	How can we prevent soil pollution? Explain.	5+5=10
	b.	What are the Negative Consequences of Soil Pollution? Explain.	
6.	a.	Define soil pollution. What are the main sources of soil pollution?	5+5=10
	b.	How does soil pollution affect soil productivity? Explain.	
7.	a.	What is Eutrophication? What are the ecological Effects of Eutrophication?	3+3+4 =10
	b.	What are the main reasons of ground water pollution?	
	c.	Write the toxic effects of arsenic and fluoride contaminated ground water. What are the remedies?	
8.	a.	What is acid rain. Write the mechanism of formation of acid rain.	3+4+3 =10
	b.	Write the causes of lead pollution. What are the harmful effects of nitrogen dioxide.	
	c.	Write a short note on Ozone layer depletion.	

[4]