D.PHARM. FIRST YEAR PHARMACEUTICAL CHEMISTRY ER20-12T JUSE OMR FOR OBJECTIVE PARTI

Duration: 3 hrs.

SET B

2024/05

Full Marks: 80

[PART-A: Objective]

Choose the correct answer from the following:				
1.	Which of the following is an example of 1st a. Cefaclor c. Cefonicid	b.	eration cephalosporins? Cefixime Cephradine	
2.	What is the name of the suffix if the ring size containing nitrogen as a hetero atom? a. irine c. inidine	b.	3 and the compound is saturatione iridine	ited
3.	Aluminium hydroxide is mainly used as- a. Haematinics c. Antacid		Protective Purgative	
4.	What is the molecular weight of Valproic ac a. 141.211g/mol c. 143.211g/mol	b.	142.211g/mol 144.211g/mol	
5.	Calcium carbonate is also known as- a. Boric acid c. Precipitated Chalk		Bleaching powder Dusting powder	
6.	pH=? a. log ₁₀ [H*] clog ₁₀ [H*]		log ₁₀ [OH·] -log ₁₀ [OH·]	
7.	The chemical name of bleaching powder is a. Calcium chloride c. Calcium oxychloride	b.	Calcium carbonate Calcium bicarbonate	
8.	Which of the following is an oxidizing agen a. Ferrous sulphate c. Stannous chloride	b.	Oxalic acid Potassium iodate	
9.	Write two examples of poorly absorbed sulpa. Sulphacetamide c. Sulphaphenazole	b.	namides? Sulphisoxazole Sulphasalazine	
10.	Which of the following is an example of 16 to a. Erythromycin c. Rokitamycin	b.	nbered natural macrolide antil Clarithromycin Josamycin	biotics?

1	 In Volhard's method, which indicator is used to confirm the prec 						
	a. Ferric alum	_	Fluorescein				
	c. Sodium rhodizonate	d.	Potassium Chromate				
13	2. When insulin was discovered?						
	a. 1921	b.	1931				
	c. 1941	d.	1951				
1.	3. Which of the following is an aprotic solv	of the following is an aprotic solvent?					
	a. Chloroform		Acetic acid				
	c. Sulphuric acid	d.	Hydrogen fluoride				
1	Which of the following is not a primary standard?						
	a. Potassium permanganate	b.	Oxalic acid				
	c. Potassium hydrogen phthalate	d.	Sodium carbonate				
1	5. Which of the following is an example of	Purine	antagonist -				
	a. 5-fluorouracil	b.	6-Mercaptopurine				
	c. Methotrexate		Imatinib				
1							
	a. 251.268g/mol		252.268 g/mol				
	c. 253.268g/mol		254.268g/mol				
1		f the following is an example of calcium channel blocker-					
1	a. Propranolol	b	Verapamil				
	c. Aliskiren		Prazosin				
1	8. Which of the following is an antidepress	Fluoxetine					
	a. Phenytoin		Phenobarbital				
	c. Alprazolam		Phenobarbital				
1							
	a. 331.745g/mol		330.745g/mol				
	c. 332.745g/mol	d.	333.745 g/mol				
20. What is the molecular formula of cyclophosphamide-							
	a. C ₇ H ₁₄ C ₁₂ N ₂ OP		C7H15C12N2O2P				
	c. C ₇ H ₁₄ C ₁₂ N ₂ O ₂ P	d.	$C_7H_{14}C_{12}N_2OP_2$				

(PART-B: Short Answers)

[Answer any ten (10) from the following]

		[3x10=30]
1.	Defiine sedatives and hypnotics? Classify them with suitable example?	1+2=3
2.	Define antianginal drugs? Classify them with suitable example?	1+2=3
3.	Define general anaesthetia? Classify them with suitable example?	1+2=3
4.	Define errors? Write two sources of error?	1+2=3
5.	Define antiviral drugs? Classify them with suitable example?	1+2=3
6.	Describe the types of solvents used for nonaqueous titration with example?	3
7.	Write down the complexing agents used in complexometric titration with structure?	3
8.	Describe the principle involved in limit test for chloride?	3
9.	What is insulin? Write a short note on regular or soluble insulin?	1+2=3
10.	Define beta-lactum antibiotics? Draw the general structure of penicillin?	1+2=3
11.	Describe structure, molecular weight and uses of Acetylcholine?	3

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